

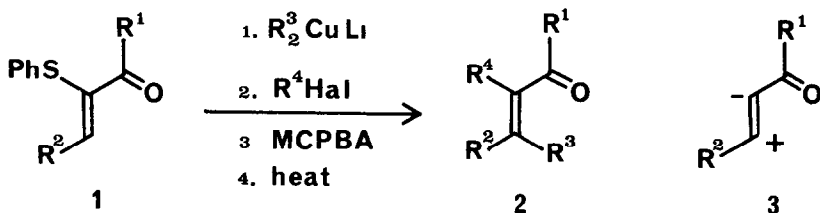
SYNTHESIS WITH MIGRATING PHENYLTHIO GROUPS
CONNECTIVE ROUTES TO 2-PHENYLTHIOALLYL
ALCOHOLS AND 2-PHENYLTHIO ENONES

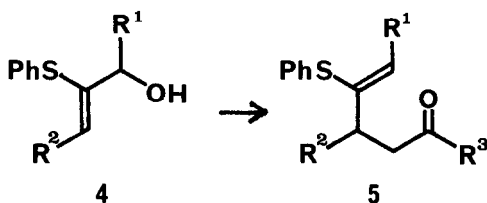
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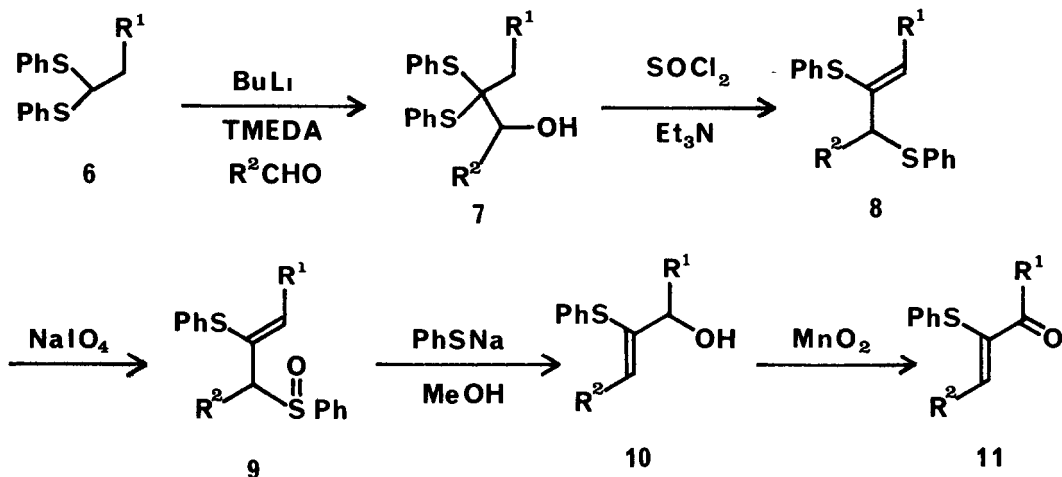
Summary The title compounds may be made from bis(phenylthio) acetals and aldehydes by routes involving PhS migration

The introduction of a phenylthio (PhS) group into a molecule often leaves unaltered the chemistry of the other functional groups but may add a latent ketone or masked double bond. Thus 2-PhS enones (1) are good Michael acceptors¹ and the double bond can be regenerated by oxidation and elimination of the PhS group (1 → 2). The enone (1) is therefore a reagent for the synthon (3). 2-PhS allyl alcohols (4) undergo normal Claisen or Carroll rearrangements² giving 1,4-dicarbonyl derivatives (5) in which the vinyl sulphide is a latent ketone.

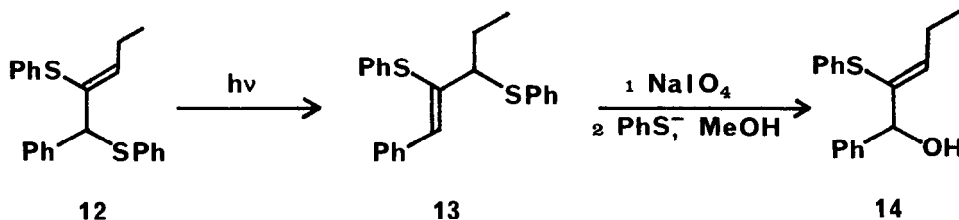




Application of these methods has so far been restricted by the absence of general routes to the enones (1) and the allyl alcohols (4). A few syntheses for specific, usually cyclic, 2-PhS enones (1) are available,³ and the alcohols (4) can be made^{2,4} by adding PhS-vinyl-lithium to carbonyl compounds but this necessarily gives (4, R² = H). We now report general, connective syntheses of both classes of compounds by routes involving [1,2] and [1,3] PhS shifts.⁵

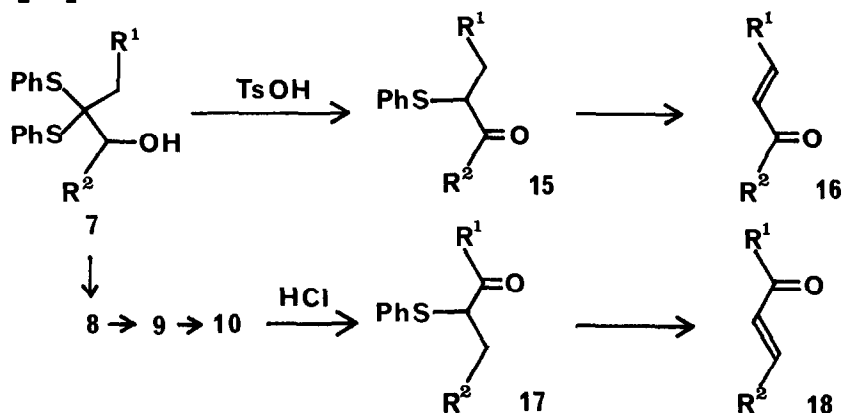


Anions from bis-PhS-acetals (6) react with aldehydes to give the adducts⁶ (7). These rearrange with PhS migration to give products^{6,7} (8) with allylic and vinylic PhS groups. The allylic PhS group can be selectively oxidised to the sulfoxide (9) which undergoes the normal [2,3] sigmatropic rearrangement⁸ in the presence of the thiophile PhS⁻ to give the 2-PhS-allyl alcohols (10). Yields are good, e.g. for R¹ = R² = Me, (7) 73%, (8) 83%, (9) 89%, and (10) 81%.



Light initiates a [1,3] PhS shift⁹ on the intermediates (8). For symmetrical compounds (8, $R^1 = R^2$) this simply alters the proportion of geometrical isomers* present. For unsymmetrical compounds (8, $R^1 \neq R^2$) the [1,3] shift may allow access to a different series. Since both the [1,3] PhS shift and the conversion of the sulphoxide to the alcohol are formal [1,3] transpositions, using them one after the other gives a product with the same skeleton as the intermediate (8). Thus (12) gives conjugated (13, 80%) in light, and hence, via the sulphoxide (79%), the alcohol (14, 75%).

We have already used⁶ the adducts (7) in a general synthesis of α -PhS-ketones (15). Cookson and Parsons⁴ report that the alcohols (10) give α -PhS-ketones (17) on treatment with acid. Two alternative series of α -PhS-ketones (15) and (17) and two series of enones (by oxidation and elimination of PhS) are therefore available from the same starting material (7). Again, the two series have a formal [1,3] relationship.



Oxidation of the 2-PhS-allyl alcohols (10) with MnO_2 in dry ether gave 2-PhS-enones (11) in moderate yield, e.g. (11, $R^1 = Et$, $R^2 = Ph$, 65%).

The reactions seem to be general except that R^1 in (8) cannot be H as the rearrangement (7 \rightarrow 8) does not give the exo-methylene compounds. The primary alcohols (10, $R^1 = H$) or enols (11, $R^1 = H$) are therefore not available by this route.

We thank S R C for a grant (to P B)

* Compounds (8-14) are formed as mixtures of geometrical isomers. We have drawn only the major isomer in each case. In the applications described here (e.g. to give 2, 3 or 17), both isomers give the same products.

References

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